

Magnetite-Platinum Core-shell Nanoparticles Using a Two-step Method for the Oxygen Reduction Reaction

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ABSTRACT

Direct Alcohol Fuel Cells (DAFCs) are a promising alternative to replace energy devices that use fossil fuels. To reach this objective, it is necessary to enhance the catalytic activity and reduce the cost of cathode catalysts. Decreasing the Pt loading on the electrode is a viable approximation to achieve lower catalyst costs. Also, the use of cheap co-catalysts helps to accomplish this goal. This is the reason for the research carried out in this work, on new nanostructures and materials as cathodes for DAFCs. It has been proved that iron oxides have a synergetic interaction with Pt to form highly active and stable anode and cathode materials. In this study, we have synthesized $\text{Fe}_3\text{O}_4@\text{Pt}$ and $\text{Fe}_3\text{O}_4@\text{Pd}$ core-shell cathode catalysts supported on Ordered Mesoporous Carbon (OMC), a novel material applied as a support on fuel cells. The synthesis has been performed in a two-step procedure. First, the formation of the ferric-ferrous oxide as core material via the reduction of the Fe precursor, using NaBH_4 as reducing agent. Second, the deposition of the Pt (or Pd) shell, followed by the dispersion of the core-shell nanoparticles on OMC, using the polyol method. The $\text{Fe}_3\text{O}_4@\text{Pt}/\text{OMC}$ and $\text{Fe}_3\text{O}_4@\text{Pd}/\text{OMC}$ catalysts have been characterized by XRD, SEM-EDS and tested as cathode materials for Oxygen Reduction Reaction (ORR) in acid medium.

Keywords: two step synthesis, core-shell nanocatalyst, Oxygen Reduction Reaction.

